THE TOTAL SYNTHESIS OF 10-(R,S)-C₃₀ BOTRYOCOCCENE AND BOTRYOCOCCANE AND A **NEW SYNTHESIS OF A GENERAL INTERMEDIATE TO THE BOTRYOCOCCENE FAMILY.**

Nicholas W. Hird, Thomas V.Lee*, Alistair J.Leigh, James R. Maxwell* and Torren M. Peakman (School of Chemistry, The University, Bristol, BS8 ITS)

Summary: The first synthesis of a C-30 botryococcene has been achieved by using a modified Julia reaction for the key *coupling offragments* (2) *and (3). The same strategy has been used in a new synthesis of a known synthetic precursor to the entire botryococcene family.*

The C₃₀-C₃₇ botryococcenes, for example the parent C₃₀ compound (1a), comprise an unusual and potentially useful family of hydrocarbon natural products found to date only in the green alga *Botryococcus* braunii¹. Interest in them stems from their biosynthesis, whereby they are thought to arise from an alternative breakdown of presqualene² (SCHEME 1), and from the potential value of the alga as a fuel source since

they can occur in remarkably high concentrations³. Furthermore, their saturated analogues, botryococcanes, show promise as biological markers in oil exploration studies⁴. Only one synthetic study of them has been reported however⁵, which is surprising since their lack of reactive functionality poses an interesting problem for their preparation.

The key to the preparation of the botryococcenes lies in the construction of the C-13 methyl group, the C-l 1 to C-12 E-alkene and the C-10 quatemary centre, which are common to all members of the family, This present paper reports on the use of a new approach to them which has resulted in the first total synthesis of lO(R,S)-C30 botryococcene **(lb).** Additionally the new route has been shown to be of use in a novel preparation of a known general intermediate to the botryococcenes which incorporates all of the key structural features

We employed a modified Julia coupling 6 for double bond formation (SCHEME 1), and the alkylation of the dienolate anion of methyl tiglate to form the quaternary centre directly⁷(SCHEME 2). The preparation of the sulphone (2) involved displacement of the iodide $(4)^8$ with the lithium anion of geranyl thiophenoxide to form the sulphide (5). Desulphurisation was best achieved using lithium in ethylamine to give the protected alcohol (6)⁹ which was converted to the sulphone (2) in 24% overall yield from the iodide (4). The aldehyde (3) was obtained by reaction of the dienolate anion of methyl tiglate with homogeranyl iodide, followed by reduction of the ester and oxidation to aldehyde (3) in 54% overall yield. Coupling of the anion of the sulphone (2) with the aldehyde (3) gave the expected hydroxysulphone (8) which after benzoylation and reduction provided the 10 (R,S) C30 botryococcene **(lb)** with the required E double bond. One diastereoisomer of this synthetic material co-eluted¹⁰ with C₃₀ botryococcene isolated from the alga, confirming it as the correct product. Additionally, exhaustive hydrogenation of **(lb)** gave 10 (R,S) **C30** botryococcane **(9)** as the only product (confirmed by mass spectrometry), showing that the only by-products produced in the synthesis are double bond isomers 11 .

The only other synthesis⁵ of a botryococcene (a C₃₄) utilised a central unit, containing the C-10 and C-13 centres common to all botryococcenes. It was of interest to see if our approach could be used in the preparation of the same unit. Thus the use of (-)-menthyl tiglate, instead of methyl tiglate, for the preparation of

i, LDA, THF, -7S", HCHO (g); ii, DHP/H+; iii, LiAlH4,THF, 76%; iv, PDC, 63%; V, **DHPm+; vi,** LiAlIQ; vii, TsCl-Py, NaI, acetone; viii, PhSO2Na. DMF, 20°, 87%; ix, n-BuLi, THF,-78O, 2Omin-(13), -78O, 30 mitt; PhCOCl, -78' to 20°, 4h, 77%; x, **Na(Hg), THF/MeOH, -15O, 55%; xi, TsOH, MeOH, 74%.**

$\rm Li$ **OTHP** $CO₂Me$ SPh (4) $\mathbf{iv}% _{1}\left(t_{1}\right) \equiv\mathbf{v}_{1}\left(t_{2}\right)$ \vert i OLi SPh OMe **OTHP** (5) v, vi ii,iii R ß, (6) $R' = OTHP$ vii $(7) R = CO₂Me$ (2) $R' = SO_2Ph$ (3) R = CHO **OH** $PhSO₂$ (8) \downarrow viii $(1b)$ \vert ix (9)

SCHEME 2

i, THF, HMPA, -78°C -RT, 3h (66%); ii, Li/NH2Et, 0°C, 1h (80%); iii, H⁺/TsCl-py/NaI, acetone/PhSO₂Na, DMF (45%); iv, LDA, THF, HMPA -78^oC; v, \sim I, THF -78^oC, 6h (77%); vi, LiAlH4, THF/PDC (54%); vii, BuLi, THF, (67%); viii BuLi, hexane -78^oC/PhCOCI/Na(Hg) MeOH/THF-20^OC (61%); ix Pd/C, H₂ (85%).

4869

the quaternary centre gave two diastereoisomers (10) , in approximately equal amounts, which are separable by chromatography. Conventional transformations then gave the aldehyde (13). The tertiary centre containing portion was prepared from (R)-methyl3-hydroxy-2-methylpropionate (11) by protection with dihydropyran and reduction with lithium aluminium hydride to an alcohol which was converted into the sulphone (12). Julia type coupling of (12) with the aldehyde (13) afforded the vicinal benzyloxy-sulphone (14). Reduction with sodium gave exclusively (15) which was deprotected to give the alcohol (16) , the key intermediate in the previous synthesis.

Acknowledgements: We are grateful to Professor E. Casadevall (Ecole Nationale Suparieure de Chemie Paris) for a gift of the hydrocarbon fraction from a culture of B. braunii and one of us (N.H.) thanks the SERC for a studentship. Acknowledgement is also made to the donors of the Petroleum Research Fund, administered by the American Chemical Society, for partial support.

REFERENCES AND NOTES

1. J.R.Maxwell, A.G.Douglas, G.Eglinton and A.McCormick, *Phytochem., 1968,2157.* R.E.Cox,

A.L.Burlingame, D.M.Wilson, G.Eglinton and J.R. Maxwell, J. *Chem. Sot. Chem. Commun. 1973,284.*

P.Metzger and E.Casadevall, *Tet. Letts.* 1983, 24, 4013. M.N.Galbraith, L.W.Hillen and L.V.Wake,

Phytochem., 1983,22, 1441. P.Metzger, E.Casadevall, M.J.Povet and Y.Povet, *Phytochem., 1985,24,2995. 2.* J.D.White, T.C.Sommers and G.N.Reddy, J. *Am. Chem. Sot., 1986,108 5352.*

3. L.V.Wake and L.W.Hillen, *Aust. J. Mar. Freshwat. Res., 1981,32,353.* A.Ben-Amotz, T.G.Tomabene and W.H.Thomas, *J. Phycol.*, 1985, 21, 72.

4. J.M.Moldowan and W.K.Seifert, /. *Chem. Sot. Chem. Commun., 1980,912.* W.K.Seifert and J.M.Moldowan, *Geochem.Cosmochim.Acta*, 1981, 45, 783. D.M.McKirdy, R.E.Cox, J.K.Volkman and V.J.Howell, *Nature*, 1986, 320, 57.

5. J.D.White, G.N.Reddy and G.O.Spessard. *J. Amer. Chem. Soc.*, **1988**, 110, 1624.

6. J.Julia and J-M.Paris, Tet. *Letts.,* 1973,4833. P.J.Kocienski, B.Lythgoe and LWaterhouse, J. *Chem. Sot. Perkins Trans.1, 1980, 1045.*

7. All yields are based on isolated and fully characterised material and all new compounds gave satisfactory analytical and/or spectroscopic data.

8. Compound (4) was prepared from (S)-methyl 3-hydroxy-2-methylpropionate by standard methods. e.g.

K.Mori, *Tetrahedron, 1983,39,3107.*

9. The alcohol (6) was contaminated with 5% of double bond isomers which were not separated.

10. GC conditions: $OV-1$ - WCOT capillary column (25 m x 0.30 mm; df $-0.17 \mu m$). Temperature programmed at 50° -300^oC at 6° C min⁻¹.

Il. The final step of the Julia sequence produced minor double bond isomers (ca. 10%) of **(lb).**

(Received in UK 21 July 1989)